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Key indicators

Single-crystal X-ray study

T = 294 K

Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$

R factor = 0.042

wR factor = 0.127

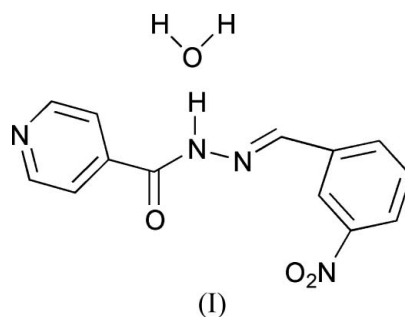
Data-to-parameter ratio = 12.0

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.3-Nitrobenzaldehyde isonicotinoylhydrazone
monohydrateThe crystal structure of the title compound, $\text{C}_{13}\text{H}_{12}\text{N}_4\text{O}_4$,
contains a solvent water molecule which interacts with the
organic molecules *via* intermolecular hydrogen bonds.

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Comment

Metal complexes incorporating Schiff bases have attracted
much attention because they can be utilized as model
compounds of active centres in various proteins and enzymes
(Kahwa *et al.*, 1986; Santos *et al.*, 2001). Their crystal structures
may provide useful information concerning their physical and
chemical properties. In this context, we report here the
synthesis and structure of the title compound, (I).The molecular structure of (I) is shown in Fig. 1. Selected
geometric parameters are listed in Table 1. The water of
crystallization interacts with the organic compound molecules
through hydrogen bonds (Fig. 2).

Experimental

An anhydrous ethanol solution (50 ml) of pyridine-4-carboxylic acid
hydrazide (1.37 g, 10 mmol) was added to an anhydrous ethanol
solution (50 ml) of 3-nitrobenzaldehyde (1.51 g, 10 mmol), and the
mixture was stirred at 350 K for 8 h under N_2 , to give a yellow
solution. The solvent was removed and the residue recrystallized
from *N,N*-dimethylformamide. The product was isolated and then
dried *in vacuo* to give pure (I) in 89% yield. Yellow single crystals
suitable for X-ray analysis were obtained by slow evaporation of an
N,N-dimethylformamide solution of (I).

Crystal data

 $\text{C}_{13}\text{H}_{10}\text{N}_4\text{O}_3 \cdot \text{H}_2\text{O}$ $M_r = 288.27$ Orthorhombic, *Pbca* $a = 13.374 (5) \text{ \AA}$ $b = 13.040 (5) \text{ \AA}$ $c = 15.742 (6) \text{ \AA}$ $V = 2745.5 (17) \text{ \AA}^3$

Z = 8

 $D_x = 1.395 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation

Cell parameters from 3032

reflections

 $\theta = 2.5\text{--}26.2^\circ$ $\mu = 0.11 \text{ mm}^{-1}$

T = 294 (2) K

Block, yellow

0.30 × 0.20 × 0.18 mm

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.969$, $T_{\max} = 0.981$
 13799 measured reflections

2430 independent reflections
 1848 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.029$
 $\theta_{\text{max}} = 25.0^\circ$
 $h = -15 \rightarrow 15$
 $k = -15 \rightarrow 15$
 $l = -11 \rightarrow 18$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.042$
 $wR(F^2) = 0.127$
 $S = 1.07$
 2430 reflections
 202 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0683P)^2 + 0.4518P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.15 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.20 \text{ e } \text{\AA}^{-3}$

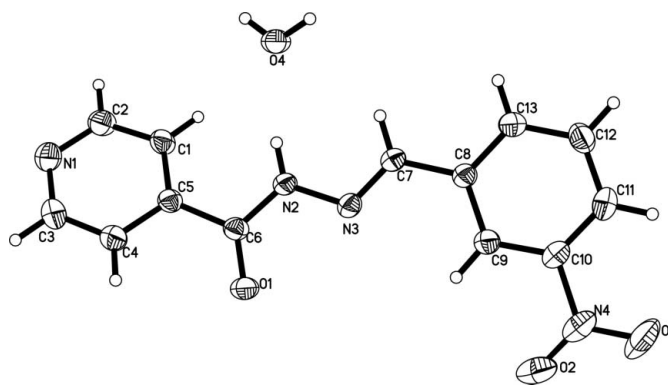


Figure 1

A view of the title compound, shown with 30% probability displacement ellipsoids.

Table 1

Selected geometric parameters (\AA , $^\circ$).

O1—C6	1.221 (2)	N3—C7	1.268 (2)
O2—N4	1.217 (2)	C5—C6	1.499 (2)
O3—N4	1.219 (2)	C7—C8	1.461 (2)
N2—C6	1.349 (2)	C8—C13	1.387 (2)
N2—N3	1.3823 (18)	C8—C9	1.391 (2)
C6—N2—N3	119.31 (13)	O1—C6—C5	121.59 (15)
C7—N3—N2	115.46 (13)	N3—C7—C8	121.14 (14)
O2—N4—O3	123.3 (2)	C13—C8—C7	120.23 (14)
O1—C6—N2	123.24 (15)	C9—C8—C7	121.16 (15)

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N2—H2 \cdots O4	0.93 (2)	1.96 (2)	2.879 (2)	170 (2)
O4—H4A \cdots O1 ⁱ	0.84 (3)	1.98 (3)	2.812 (2)	168 (2)
O4—H4B \cdots N1 ⁱⁱ	0.86 (3)	2.01 (3)	2.876 (2)	176 (2)
C1—H1 \cdots O4	0.93	2.56	3.282 (3)	134
C7—H7 \cdots O4	0.93	2.58	3.360 (2)	142
C12—H12 \cdots O3 ⁱⁱⁱ	0.93	2.39	3.221 (3)	148

Symmetry codes: (i) $x + \frac{1}{2}, y, -z + \frac{1}{2}$; (ii) $-x + \frac{3}{2}, -y + 1, z + \frac{1}{2}$; (iii) $x + \frac{1}{2}, y, -z + \frac{3}{2}$.

All C-bound H atoms were positioned geometrically and refined using the riding-model approximation, with $C-H = 0.93 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The amino and water H atoms were located in a difference Fourier map and refined freely (Table 2).

Data collection: SMART (Bruker, 1999); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve

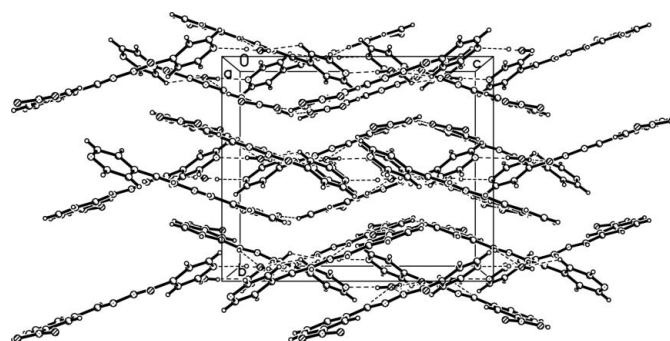


Figure 2

A view down the a axis of the crystal structure, showing the extensive intermolecular hydrogen-bonding interactions (dashed lines).

structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

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